TOOTH PASTES

A toothpaste or dentifrice is a substance used with a toothbrush for the purpose of cleaning the accessible surfaces of the teeth.

Purposes:

- Cleaning
- Polishing
- Removal of stains
- Reduce incidence of tooth decay
- Reduction of oral malodors



- Delivers active ingredients such as fluoride or xylitol to help prevent tooth and gum disease.
- Recent advances in toothpaste enable high efficacy of oral health delivery:
 - Special toothpaste for kids with fluoride and relatively low abrasion value
- Desensitizing toothpaste
- Whitening toothpaste
- Easy to use, available in collapsible tubes.
- Combination of various contents available as to combat different oral healthcare needs at once.

DISADVANTAGES

- Extended consumption of large volumes of fluoridated toothpaste can result in fluorosis.
- Fluoridated toothpaste can be either acutely toxic if swallowed in large amounts or chronically toxic if swallowed in any amount consistently.
- Triclosan, an active ingredient in many toothpastes can combine with chlorine in tap water to form chloroform which is a human carcinogen. According to some scientists it can leave unborn babies brain damaged.
- Although in several studies whitening toothpaste show the ability to improve tooth color they have side effects.
- The most significant one is enamel and dentin abrasion which in turn leads to increased tooth sensitivity.

TOOTHPASTE FORMULATION

General toothpaste formulation composition contains:

- 1. Abrasives
- 2. Detergents
- 3. Water
- 4. Humectants
- 5. Thickening Agents
- 6. Flavor
- 7. Sweeteners8. Preservatives
- 9. Corrosion Inhibitors
- 10. Colorants
- 11. Bleaches
- 12. Anticaries Actives
- 13. Anticalculus Agents
- 14. Desensitizing Agents
- 15.Antimicrobials /antiplaque/ antigengivitis agents

1. Abrasives:

- The main component responsible for cleaning the teeth.
- Should be abrasive enough to clean the tooth & avoid damage to tooth surface
- Ability depends on particles size, shape, & brittleness of the material.

EXAMPLES:

- →hydrated silica
- \rightarrow calcium carbonate,
- →calcium pyrophosphate
- \rightarrow dicalcium phosphate dihydrate

2. Detergents:

major contributor to teeth cleaning process

Should be tasteless, nontoxic, nonirritant, producing large volume of foam.

Examples:

 \rightarrow Sodium lauryl sulfate (SLS),

→Sodium lauryl sarcosinate

3. Humectants :

Prevents toothpaste from drying outGlycerin (50% aq. solution)

EXAMPLES:

 \rightarrow Sorbitol

 \rightarrow hydrogenated starch hydrolysate

 \rightarrow Propylene glycol, PEG :

Disadvantage: bitter taste

Advantage : 1. sweet & cooliness feeling

2. anticaries

4. Thickening Agents:

- Needed to maintain the stability of high-solid dispersion
- Affect the dispersibility ,foam character, and mouth feel

Examples:

→Sodium carboxymethylcellulose (interacts with cationic antimicrobials),
→Hydroxypropyl methyl cellulose (HPMC)
→ hydroxyethyl cellulose (nonionic),
→Xanthan gum,
→carbomer

5. Flavour

- Influence consumer acceptance
- Leaves a fresh clean feeling after brushing
- Considered highest cost ingredient in the toothpaste formulation

Examples : Peppermint, spearmint, menthol, wintergreen, cinnamon,...

6. Sweeteners :

- Most flavor oils have bitter taste leaving the toothpaste formulation unpalatable without sweeteners
- Artificial sweeteners are used (non cariogenic effect like natural sugar)

Examples :

- →Saccharin, sodium saccharin
- \rightarrow potassium acesulfame

7. Preservatives :

Rarely used nowadays due to safety concerns

Proper formulation ingredient selection can result in preservative free toothpaste

8. Corrosion inhibitors: :

- Used to avoid corrosion of the aluminum tubes (e.g. sodium silicate)
- Not needed nowadays due to the use of plastic tubes

9. Anticaries Actives:

Fluoride ions reduce the incidence of carious lesion by reducing the acid solubility of tooth enamel

Examples :

- → Sodium fluoride
 → sodium monofluorophosphate
- \rightarrow stannous fluoride

FDA recommends levels of soluble fluoride ion between 850-1150ppm

10. Anticalculus Agents:

Examples:

 \rightarrow zinc chloride,

 \rightarrow zinc citrate (Crest)

 \rightarrow tetra sodium pyrophosphate,

→ disodium pyrophosphate (used more than zinc salts, and it has antiplaque activity)

11. Desensitizing Agent :

- Potassium Nitrate(5%)
- Compatible with fluoride, and has salty taste
- Ex. Sensodyne toothpastes

12. Antibacterial/Antiplaque/antigingivitis:

Triclosan, Chlorhexidine, Zinc citrate.

GENERAL METHOD OF PREPARATION

- The preparations are preferably made in stainless steel mixer container for large scale manufacture filled with slowly rotating blades.
- It can be done for in planetary mixer or similar mixer used for semisolid preparations. Small scale batch can be made in glass container.
- The gum is mixed with suitable quantity of humectant without any water proper dispersion.
- Chloroform and talc. Can also be used for dispersion of binding agent.
 Other colloid may be disperse in water.
- Methyl cellulose should be mixed with cold water, but ethyl cellulose should be disperse in warm water.
- Other powder ingredient are sifted together and added gradually to mucilaginous mixture with continue Gentle stirring.

Then aq. Media is mixed and stirred to get a product.

Flavour and detergent added at the last.

Formula 1	
Calcium carbonate	56.0 g
Na.lauryl sulphate	1.0 g
Glycerin	22.0 g
Gum tragacanth	1.5 g
Water	19.4 g
Saccharine	0.1 g
Flavour	q.s.
Preservative	q.s.

Anti-cavity Agents:

(a) Sodium Monofluorophosphate (Na2PO3F)
(b) Sodium Fluoride (NaF)
(c) Organo (amine) Fluorides
(d) Stannous Fluoride (SnF2)











EVALUATION OF TOOTHPASTE

- 1. Test for abrasive character
- 2. Determination of particle size
- 3. / Determination of cleansing property
- 4. Consistency
- 5. pH determination
- 6. Foaming power
- 7. Determination of fluoride ion
- 8. Stability
- 9. Determination of moisture and volatile matter
- 10. Limit test for heavy metal.